



Nuclear Science and Security Consortium
Virtual Scholar Showcase 2020

**Synthesis of Li-containing Garnet Ceramics for
Dual-mode Detection**

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SCINTILLATION MATERIALS
RESEARCH CENTER



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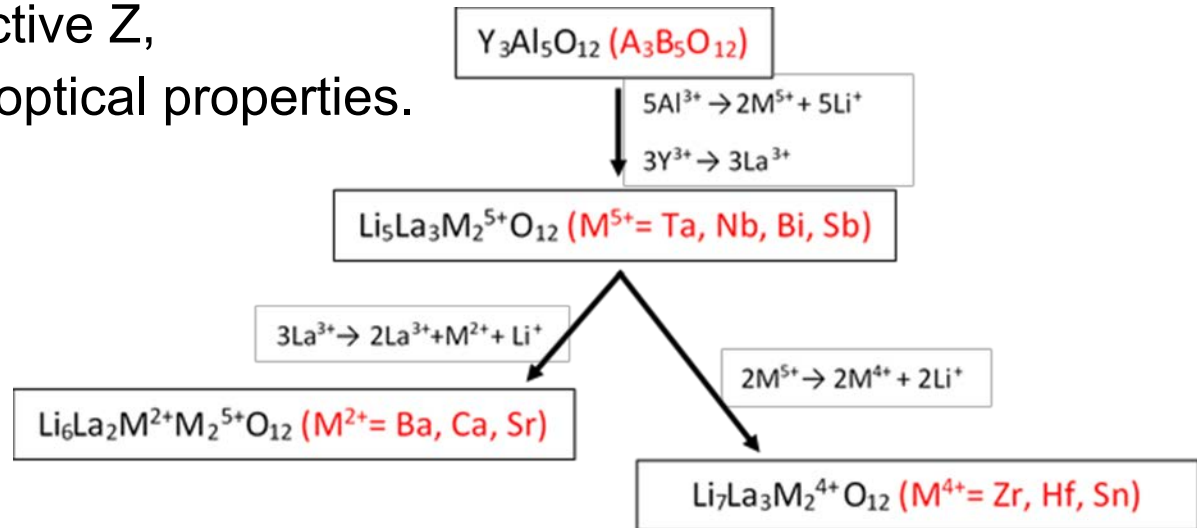
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(Sigma Division)



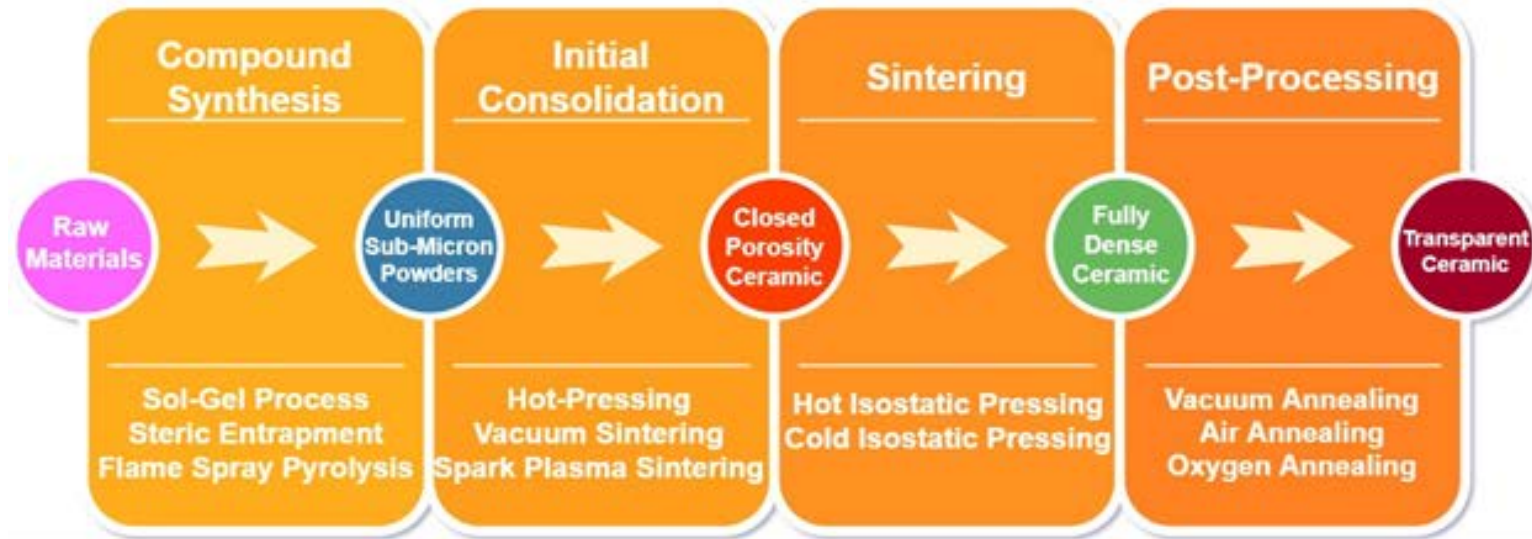
NSSC Focus Area: Nuclear Instrumentation
Radiation Detection and Instrumentation

- One of the goals of the NNSA is to **“develop technologies to detect nuclear and radiological proliferation worldwide”**.
- Wide spread use of novel dual-mode inorganic scintillators such as LiI:Eu and $\text{Cs}_2\text{LiYCl}_6\text{:Ce}$ (CLYC) has been prevented by their high cost and low yields from current single crystal growth methods
- Transparent ceramic scintillators could provide a quicker and cheaper method of producing high performance scintillators in near net shapes

- Initially researched for solid state electrolytes for Li-ion batteries due to high Li-content and garnet structure
- Among these, $\text{Li}_{7-x}\text{La}_3\text{Zr}_{2-x}\text{Ta}_x\text{O}_{12}$ (LLTZx) garnets are promising for efficient dual mode ceramic detectors due to their:
 - high Li content,
 - high density,
 - high effective Z,
 - isotropic optical properties.



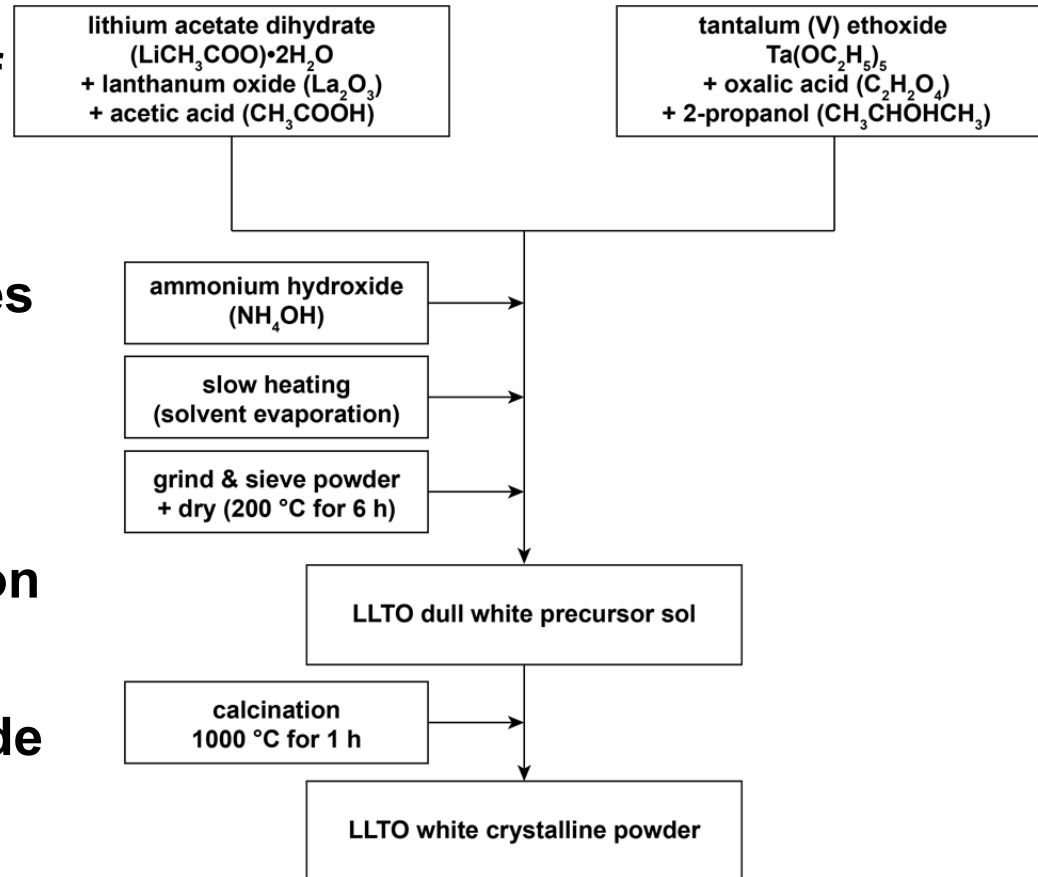
A Process for the Synthesis of Transparent Ceramics



- **A general process for synthesizing transparent ceramics**
- **Several challenges include:**
 - Secondary phases from volatilization of Li during high temperature stages
 - Intragranular pores from imbalance of grain growth and pore removal

Compound Synthesis via Sol-gel Process

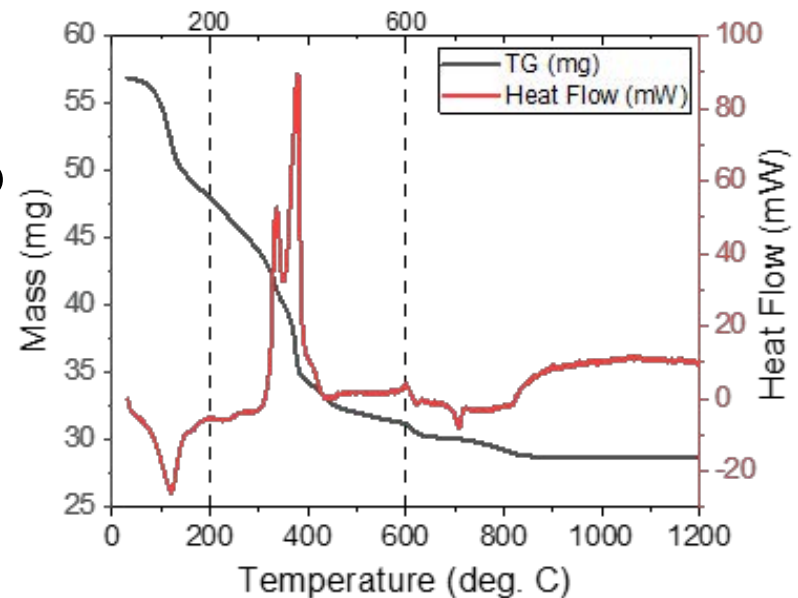
- A flexible alkoxide sol-gel process developed with LANL for the synthesis of $\text{Li}_5\text{La}_3\text{Ta}_2\text{O}_{12}$ (LLT)
- Solubility of Li-intermediates necessitates solvent evaporation
- Li-volatility at high temperatures requires compensation by inclusion of excess Li-acetate
- Substitution of Ta-ethoxide for Zr-isopropoxide for LLTZx



Establishing Boundary Conditions for Thermal Processing

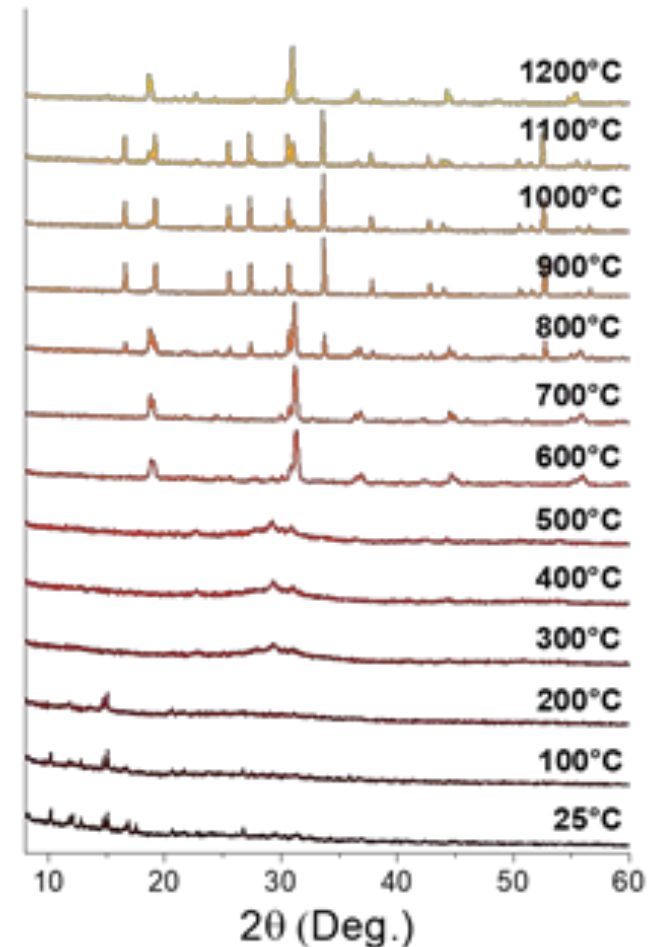
Simultaneous DSC/TGA of dried sol-gel cake to better understand transformation to crystallized oxide

- **20°C- 200°C**
 - Removal of residual water and volatile organics
- **200°C- 600°C**
 - Decomposition of oxalates and acetates to carbonates/oxycarbonates
- **600°C and up**
 - Decomposition of carbonates/oxycarbonates, crystallization and glass transition related to reconfiguration



Equilibrium High Temperature X-Ray Diffraction of dried sol-gel cake

- **20°C- 200°C**
 - Low angle peaks likely from residual organics
- **200°C- 600°C**
 - Low intensity peaks around 30° are likely from carbonates/oxycarbonates
- **600°C and up**
 - $\text{LiLa}_2\text{TaO}_6$ phase first appears, then transitions into $\text{Li}_5\text{La}_3\text{Ta}_2\text{O}_{12}$ which then begins to break down from Li-volatilization



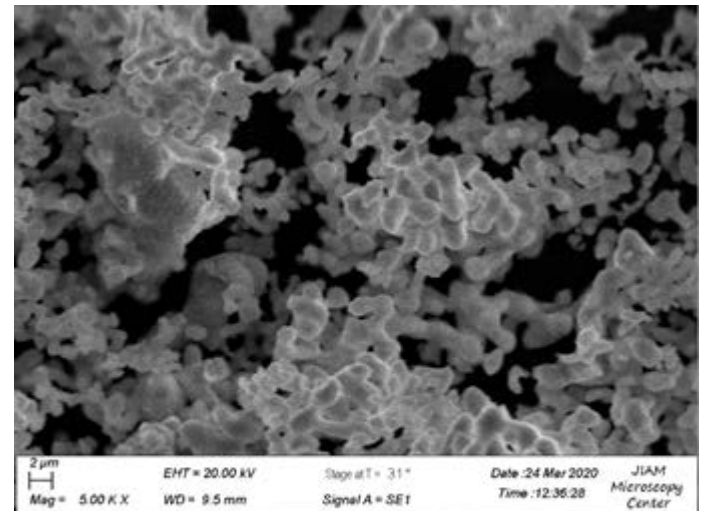
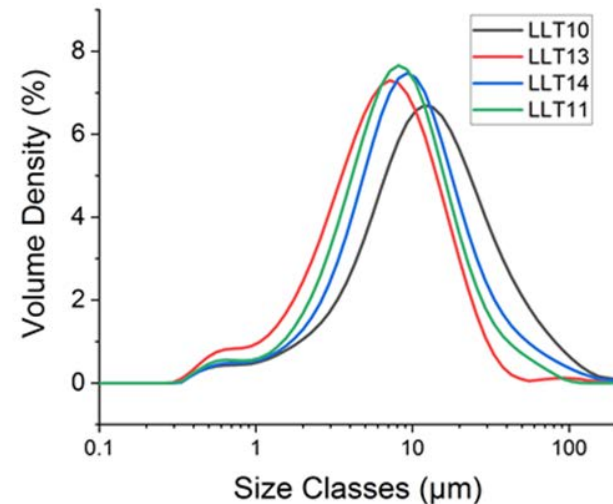
Li-volatility during crystallizing and initial consolidation necessitates excess Li to compensate

- **3 batches of powder were made with varying amounts of excess Li following the same sol-gel procedure**
- **1 batch was made with excess La**

Sample	Wt% Excess Li	Excess La
LLT10	10	No
LLT13	12.5	No
LLT14	12.5	1 Wt%
LLT11	15	No

Particle size distribution was analyzed for each batch of powder using Dynamic Light Scattering (DLS) and Scanning Electron Microscopy (SEM)

- **Uniform size distribution of each of the powder batches with similar mean size**
- **Observation using SEM shows sintering of particles during crystallization resulting in larger agglomerates.**



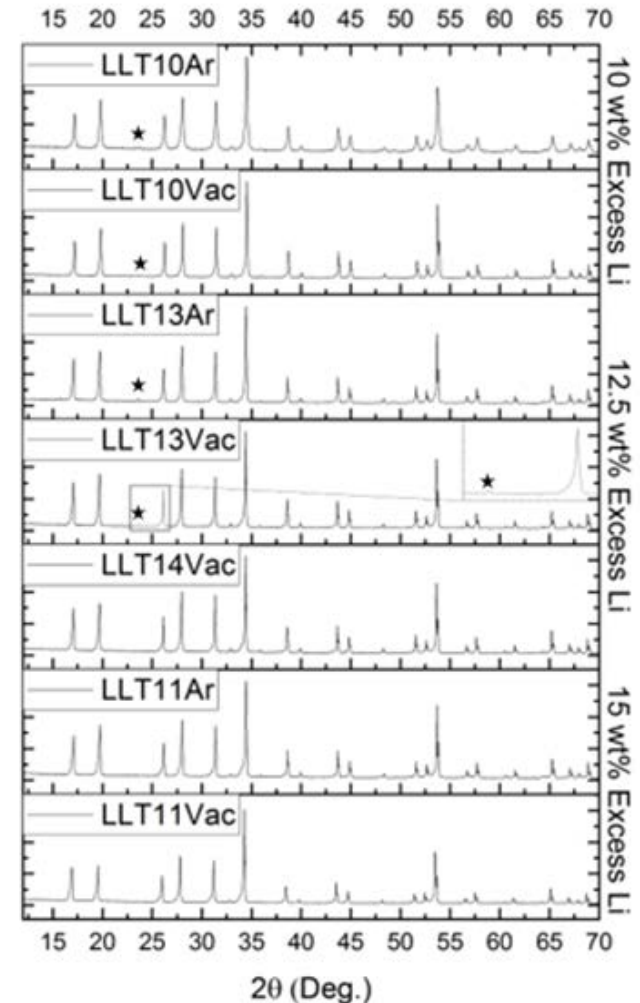
- ~4.5 g green bodies at 182 MPa in a 12 mm diameter stainless steel die
- Green bodies are then loaded into a 12 mm diameter grafoil-lined graphite die and hot pressed at 1200 °C and 80 MPa for 1 hr under vacuum or flowing Ar.

Sample	Wt% Excess Li	Excess La	Hot Pressing Atmosphere	Density [g/cm ³]	Secondary Phases
LLT10Ar	10	No	Ar	6.17 (97.5% TD)	LiTaO ₃ , Li ₇ La ₃ Ta ₂ O ₁₃
LLT10Vac	10	No	Vacuum	6.18 (97.6% TD)	LiTaO ₃ , Li ₇ La ₃ Ta ₂ O ₁₃
LLT13Ar	12.5	No	Ar	6.28 (99.2% TD)	LiTaO ₃ , Li ₇ La ₃ Ta ₂ O ₁₃
LLT13Vac	12.5	No	Vacuum	6.32 (99.8% TD)	LiTaO ₃ , Li ₇ La ₃ Ta ₂ O ₁₃
LLT14Vac	12.5	1 Wt%	Vacuum	6.15 (97.2% TD)	Li ₇ La ₃ Ta ₂ O ₁₃
LLT11Ar	15	No	Ar	6.12 (96.7% TD)	Li ₇ La ₃ Ta ₂ O ₁₃
LLT11Vac	15	No	Vacuum	6.20 (97.9% TD)	Li ₇ La ₃ Ta ₂ O ₁₃

Secondary Phase Formation and the Effect of Excess Li

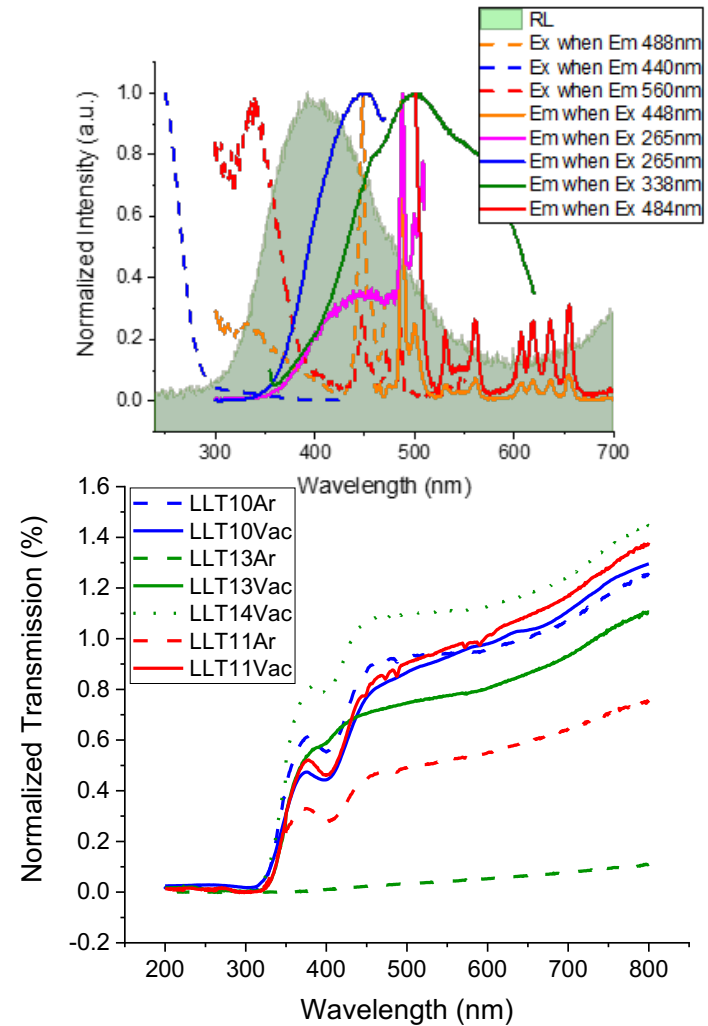
X-Ray Diffraction of hot pressed samples show near single phase ceramics.

- **Low intensity peak at $\sim 23^\circ$ from LiTaO_3 secondary phase**
- **Garnet peaks show low angle asymmetry from formation of Li-stuffed $\text{Li}_7\text{La}_3\text{Ta}_2\text{O}_{13}$ garnet phase**
- **Rietveld refinements using GSAS II resulted in poor fits from asymmetry of garnet peaks**



Broad X-Ray excited radioluminescence and photoluminescence emission from self-trapped exciton tantalate intrinsic emission

- Multiple line emissions above 480 nm likely from a 4f-4f transition in some rare-earth impurity.
- In-line transmission spectra show a degree of transparency in the area of emission



- **Hot isostatic pressing of consolidated ceramics**
- **Expand sol-gel process to include Zr**
- **Investigate activation by Ce^{3+} and Pr^{3+} using the sol-gel process**
- **Investigate pulse shape and pulse height discrimination techniques for neutron-gamma discrimination**

- **LANL School of Nukes 2018**
 - Non-destructive assay short course
- **NSSC-LANL Keepin Non-proliferation Summer Program 2018**
 - Resulted in a continued collaboration on the synthesis of nano-powders and transparent ceramics with Dr. Chen (LANL)
- **GW Bootcamp on Nuclear Security Policy 2019**
 - An in-depth introduction to Nuclear Security policy and job opportunities
- **Several Conferences and Presentation Opportunities**



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