







Experimental Investigation of a Machine Learning Model for Scintillator Discovery

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Introduction



Department and University: Materials Science and Engineering
Department, University of Tennessee
Academic Advisor: Dr. Mariya Zhuravleva
NSSC Research Focus Areas: Radiation Detection and Instrumentation
Planned Graduation Date: Dec. 2023

Lab Mentor/Partner Laboratory: Dr. Kenneth McClellan (LANL, MST-8)

Research Description: Discovery and development of novel scintillator materials



High Energy Resolution – Cs, Srl,:Eu

Fast timing – Cs_2ZnCl_4 and Cs_3ZnCl_5

ML Prediction Model – Double Perovskites



- 1-2 ns decay times
- Coincidence time resolution (CTR) as good as 137 ps FWHM
- Able to grow in large sizes



LANL internship Topic of today's talk

- 3.0% ER at 662 keV for small crystals
- Comparable performance to Srl₂:Eu
- Able to grow in large sizes

Relevance to NSSC / NNSA Mission

Mission: reduce global nuclear security threats through the innovation of technical capabilities to *detect*, *identify*, and *characterize* illicit diversion of special nuclear materials

The focus of this research is on developing scintillators with better capabilities (energy resolution, decay time, light yield, etc.) for nuclear safeguards and nonproliferation applications

• Ex. – use in handheld RIIDs, X-ray imaging systems, etc.



Experimental Investigation of Scintillator Prediction Model LANL Internship Project

 Machine learning (ML) model has been developed for high-throughput screening of potential new scintillator materials

Physics-informed machine learning for inorganic scintillator discovery

Cite as: J. Chem. Phys. **148**, 241729 (2018); https://doi.org/10.1063/1.5025819 Submitted: 13 February 2018 . Accepted: 05 April 2018 . Published Online: 25 April 2018

🝺 G. Pilania, K. J. McClellan, C. R. Stanek, and 匝 B. P. Uberuaga

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- Objectives of this work
 - Provide experimental data to feed back into model
 - Identify promising scintillator compounds to pursue further
- Motivation:
 - Ideal scintillator does not exist
 - Discovery of new scintillator compounds can be lengthy and tedious
 - Complimentary use of modeling and experimental work can provide a more efficient approach than conventional "trial and error"
 - Rapid screening, vast chemical spaces

Project Overview

- Compositional space double perovskite halides A₂BB'X₆:Ce (A = Rb⁺, Cs⁺; B = Na⁺, K⁺; B' = RE³⁺; X = Cl⁻, Br⁻)
- Selected a set of *new compounds* to pursue experimentally, based on calculated electronic band structures
- Fabrication method single crystal growth via Bridgman technique
- Characterization physical, optical, and scintillation properties



Double perovskite structural

Cubic space group *Fm*-3*m*

Positions of Ce^{3+} 5*d* and 4*f* energy levels relative to CB and VB for 200 Double Perovskite Halides



ML model predicts that some of these will be efficient scintillators

G. Pilania *et al.* "Physics-informed machine learning for inorganic scintillator discovery" J. Chem. Phys. **148**, 241729 (2018)

Initial Screening Done w/ RbCI-based Compounds

-	Composition	Calc. Density (g/cm3)	Z _{eff}
3 of the predicted scintillators, based on calculated positions of Ce ³⁺ energy levels	Rb ₂ NaLaCl ₆ :Ce 2%	2.67	39.1
	Rb, NaGdCl,:Ce 2%	2.93	43.5
	Rb ₂ NaYCl ₆ :Če 2%	2.61	30.7
Screening process Set up equipment and capabilities	Synthesize evaluate polycrystal samples	and line Grow sin crysta	ngle Is ion) k into prediction models

Setting Up Equipment and Capabilities

(1) Vacuum Drying Station



Raw material is dried under vacuum (~10⁻⁵ mbar) to remove moisture / hydrates



Encapsulated and dried powder, ready for synthesis

Lab was previously equipped for growth of *oxides*, not *halides*; require different handling and synthesis/growth methods.



(2) Growth Furnace

Successfully Synthesized Rb₂NaYCl₆:Ce and Rb₂NaGdCl₆:Ce

- XRD shows primary phase is cubic elpasolite (space group Fm-3m)
- Low intensity peaks from secondary phase(s) also observed in XRD
 - Match with NaCl and Rb₃(RE)Cl₆ phases
 - Expected due to incongruent melting

Synthesized Polycrystalline Samples

- Rb₂NaLaCl₆:Ce appears to be unstable compound
 - Instead, decomposes into Rb₂LaCl₅ and NaCl
 - Results confirmed w/ second sample prepared at UT



Measured XRD of Polycrystalline Samples vs Reference Patterns

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2/3 double perovskite compounds correctly predicted



Measured XRD of Polycrystalline Samples vs Reference Patterns

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Photoluminescence Measurements Indicate Successful Incorporation of Ce³⁺ Activator

- Measured photoluminescence (PL) of the polycrystalline samples to assess which are promising
 - Uses *light* as excitation source
 - If no emission, could suggest unfavorable Ce³⁺ energy levels
- Strong doublet emission is observed from Ce³⁺ 5d to 4f electronic transition
- Based on result, proceeded with growth



Photoluminescence Spectra



Growth Yielded Transparent Single Crystals of Rb₂NaYCl₆:Ce and Rb₂NaGdCl₆:Ce



XRD of Different Sections of Crystal



- Middle section of each crystal is transparent
- Phase purity analyzed w/ XRD
 - Shows middle is nearly phase pure for both
 - 3 distinct sections observed, common for elpasolites grown by directional solidification

XRD of Different Sections of Crystal



Both Crystals Scintillate Under X-ray and Gamma Excitation



Decay Constants

	T ₁ (ns)	T ₂ (ns)	Τ ₃ (μs)
Rb ₂ NaGdCl ₆ :Ce	94 (2%)	632 (22%)	3.72 (76%)
Rb ₂ NaYCl ₆ :Ce	85 (25%)	340 (56%)	1.78 (19%)

Rb₂**NaGdCl**₆:Ce Has Moderate Light Yield and Energy Resolution

- Scintillation properties further evaluated by pulse height analysis
 - performed on 8 mm diameter slabs, ~3-4 mm thick
- Rb₂NaGdCl₆:Ce has LY of 14,000 ph/MeV
- Energy resolution is 5.5% at 662 keV
- Rb₂NaYCl₆:Ce has LY of 4,400 ph/MeV



 $\text{ER} = \frac{FWHM}{photopeak \ centroid}$



^{*}Gd escape peak at 43 keV lower than the full energy photopeak

Summary and Conclusions

- Single crystals of Rb₂NaYCl₆:Ce and Rb₂NaGdCl₆:Ce were grown
- Both were correctly predicted to scintillate
- Generated the first set of experimental data for a ML-based model for scintillator discovery
- Capabilities now set up for MST-8 group at LANL to continue work on synthesis and growth of novel halide scintillators

Although there is a wide range of compositions that will eventually need to be studied experimentally, this work is an important first step toward validating the ML model developed by Pilania *et al.*

The NSSC Experience

NSSC sponsored events

- Presented at UPR (2018, 2019)
- Presented at NSSC Fall Workshop (LLNL 2019)
- NSSC-LANL Keepin Nonproliferation Science Summer Program (2021)

Other conferences, workshops, etc. attended w/ funding from NSSC

- PNNL Radiation Detection for Nuclear Security Summer School (2018)
- International Summer School on Crystal Growth (ISSCG-17) in Granby, CO (2019)
- Presented at several conferences: SORMA (2018), ACCGE (2017, 2021), ICCGE (2019), IEEE (2020, 2021)

Connection with national labs

- Conducted summer research at LANL (2021)
- Spent 2 weeks at LLNL performing measurements (2019)
- Still collaborating w/ group at LANL



Plaza Blanca (Abiquiu, NM)

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Questions?